

# Read PDF Balancing Chemical Equations

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The first step to balance  
the equation is to write  
down the chemical formula of  
reactants that are listed on  
the left side of the

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Answers Key  
chemical equation. After this, you can list down the products on the right hand side of the chemical equation. There is an arrow between the sides, signaling the direction the reaction is happening in.

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## *Balancing Chemical Equations*

...

Balancing Chemical Equations

– Answer Key Balance the equations below:

- 1)  $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$
- 2)  $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$
- 3)  $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$
- 4)  $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$
- 5)  $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$
- 6)  $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$
- 7)  $1 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow 1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$
- 8)  $1 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$
- 9)  $2 \text{ C}_8\text{H}_{18} + 25 \text{ O}_2 \rightarrow 16 \text{ CO}_2 + 18 \text{ H}_2\text{O}$
- 10)  $1 \text{ FeCl}_3 + 3 \text{ NaOH} \rightarrow 1 \text{ Fe(OH)}_3 + 3 \text{ NaCl}$
- 11)  $4 \text{ P} + 5 \text{ O}_2 \rightarrow 2 \text{ P}_2\text{O}_5$

*Balancing Chemical Equations*

To set up an equation in the Chemical Equations Gizmo,

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Answers Key  
type the chemical formulas into the text boxes of the Gizmo. First, type in "H<sub>2</sub>+O<sub>2</sub>" in the Reactants box and "H<sub>2</sub>O" in the Products box. This...

## *Student Exploration- Balancing Chemical Equations (ANSWER KEY)*

Balancing Equation And Answer Key - Displaying top 8 worksheets found for this concept.. Some of the worksheets for this concept are Balancing equations practice problems, Balancing chemical equations work and answer key, Balancing equations work and key 7 23 09, Balancing equations work answers, Balancing equations



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A chemical equation is  
balanced by changing the  
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coefficients LIGHT BLUE  
anything necessary DARK  
GREEN 10 What is the mass of  
Na?  $2\text{Na} + 3\text{N}_2 \rightarrow 2\text{Na} + 3\text{N}_2$   
(500g) (?) (323g) 100 grams  
ORANGE 177 grams PINK 323  
grams YELLOW

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Answers Key  
Chemical Equations and click on the first Link or go to <https://phet.colorado.edu/en/simulation/balancing-chemical-equations> Click on the "Run in HTML 5" BUTTON  
PRE-LAB: Use the simulation to adjust the coefficients and balance the following equation. Remember a coefficient cannot be zero.

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replacing names with  
formulae, it may not be  
balanced. The numbers of  
atoms of each element on the  
left must be the same as  
they are on the...

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Anne Marie Helmenstine,  
Ph.D. Updated January 04,  
2019. A balanced chemical  
equation gives the number  
and type of atoms  
participating in a reaction,  
the reactants, products, and  
direction of the reaction.  
Balancing an unbalanced  
equation is mostly a matter  
of making certain mass and  
charge are balanced on the  
reactants and products side  
of the reaction arrow.

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*How to Balance Equations -  
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The must find the equation that corresponds to what is inside each atom and then follow the arrow with the correct answer to the next atom. Once an atom is used it cannot be revisited, even if an arrow with the correct answer points back to it, the student must find another path and make there way to the end.

*Science Maze Balancing  
Chemical Equations by  
Science of ...*

Balancing Equations About  
Chemistry

<http://chemistry.about.com>



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Answers Key  
Balance the following  
chemical equations. 1.  $\text{Fe} + 2\text{H}_2\text{SO}_4 \rightarrow \text{FeSO}_4 + \text{H}_2$   
2.  $\text{C} + \text{H}_2\text{O} \rightarrow \text{CO} + \text{H}_2$   
3.  $\text{KOH} + \text{H}_3\text{PO}_4 \rightarrow \text{K}_3\text{PO}_4 + \text{H}_2\text{O}$   
4.  $\text{SnO}_2 + \text{H}_2 \rightarrow \text{Sn} + \text{H}_2\text{O}$   
5.  $\text{NH}_3 + \text{O}_2 \rightarrow \text{NO} + \text{H}_2\text{O}$   
6.  $\text{KNO}_3 \rightarrow \text{K}_2\text{CO}_3 + \text{HNO}_3$   
7.  $\text{B} + \text{Br}_2 \rightarrow \text{HNO}_3$   
8.  $\text{BF}_3 + \text{Li}_2\text{SO}_3 \rightarrow \text{Li}_2\text{SO}_3 + \text{H}_2\text{O}$

Name: \_\_\_\_\_ Date: \_\_\_\_\_  
Balancing Equations

Balancing Equations: Answers  
to Practice Problems. 1.  
Balanced equations.

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(Coefficients equal to one (1) do not need to be shown in your answers). (a)  $2 \text{Fe} + 3 \text{Cl}_2 \rightarrow 2 \text{FeCl}_3$ . (b)  $4 \text{Fe} + 3 \text{O}_2 \rightarrow 2 \text{Fe}_2\text{O}_3$ . (c)  $2 \text{FeBr}_3 + 3 \text{H}_2\text{SO}_4 \rightarrow 1 \text{Fe}_2\text{SO}_4$ .

## *Chemical Equations Answers - Teacher Worksheets*

There are 10 chemical equations for students to balance. Each equation comes with a helpful chart so students practice counting up the number of atoms on each side and making sure they match. This worksheet is perfect for a student learning how to balance an equation for the first time! Give as classwork, assign as homework, or leave for a

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