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C11H22 + O2 21. aluminum

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chloride + potassium sulfide

22. sodium sulfite +

hydrochloric acid ch3

chemical reactions 5/19/2020

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CHEMICAL EQUATIONS For each

of the following equations

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write a formula equation.

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Reactions In decomposition
reactions, one compound will

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break down into two or more parts.

1. barium carbonate \rightarrow barium oxide + carbon dioxide
 $\text{BaCO}_3 \rightarrow \text{BaO} + \text{CO}_2$

2. magnesium carbonate \rightarrow magnesium oxide + carbon dioxide
 $\text{MgCO}_3 \rightarrow \text{MgO} + \text{CO}_2$

3.

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difficulty in balancing
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Balancing an unbalanced equation is mostly a matter of making certain mass and charge are balanced on the reactants and products side of the reaction arrow. This is a collection of printable

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Worksheet 11.3.2 is identical to Equation 11.3.1 except for the addition of abbreviations in parentheses to indicate the physical state of each species. The abbreviations are (s) for solid, (l) for liquid, (g)

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for gas, and (aq) for an aqueous solution, a solution of the substance in water.. Consistent with the law of conservation of mass, the numbers of each type of atom are the same on both ...

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AP Chemistry Chapter 4

Worksheet # 2 Name _____ 1.

Write equations showing the
dissociation of a) NaCl b)
MgCl₂ c) PbCl₂. a) b) c) 2.

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Write equations showing the ionization of a) HNO_3 b) HNO_2 . b) 3. Write equations showing the dissociation of a) $\text{Ca}(\text{OH})_2$ b) $\text{Mg}(\text{OH})_2$. b) 4.

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